Model Answer

Candidates answer on the Question Paper.
No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.
Write in dark blue or black pen.
You may use a pencil for any diagrams, graphs or rough working.
Do not use staples, paper clips, highlighters, glue or correction fluid.
DO NOT WRITE IN ANY BARCODES.

Answer all questions.

At the end of the examination, fasten all your work securely together.
The number of marks is given in brackets [ ] at the end of each question or part question.

This document consists of 12 printed pages.
An IGCSE student is investigating the passage of light through a transparent block using optics pins.

The student's ray trace sheet is shown in Fig. 1.1.

The student places two pins $P_1$ and $P_2$ to mark the incident ray. He looks through the block and places two pins $P_3$ and $P_4$ to mark the emergent ray so that $P_3$, $P_4$ and the images of $P_1$ and $P_2$ appear to be exactly one behind the other. He draws the outline of the block. He removes the block and pins and draws in the incident ray and the emergent ray.

(a) (i) On Fig. 1.1, mark suitable positions for the four pins. Label the pins $P_1$, $P_2$, $P_3$ and $P_4$.

(ii) Draw the normal at point $A$. 

[2]
(b) (i) Draw in the line $AB$. Measure and record the angle of refraction $r$ between the line $AB$ and the normal.

$$r = 18^\circ \rightarrow 22^\circ$$

(ii) Measure and record the angle of incidence $i$ between the incident ray and the normal.

$$i = 30^\circ \rightarrow 34^\circ$$

[Co-unit is needed]  

(c) The student does not have a set square or any other means to check that the pins are vertical. Suggest how he can ensure that his $P_3$ and $P_4$ positions are as accurate as possible.

[View bases of pins]  

(Keep line of sight low / View close to table)  

[Total: 5]
An IGCSE student is investigating the energy changes that occur when hot water and cold water are mixed.

The student is provided with a supply of hot water and a supply of cold water.

The temperature of the cold water \( \theta_c = 23^\circ C \).

(a) The temperature of the hot water is shown in Fig. 2.1.

![Fig. 2.1](image)

Record the temperature \( \theta_h \) of this hot water.

\[ \theta_h = 83^\circ C \] [1]

(b) The student pours 50 cm\(^3\) of the hot water into 50 cm\(^3\) of the cold water. He briefly stirs the mixture and then records the temperature \( \theta_m \) of the mixture, \( \theta_m = 49^\circ C \).

(i) Calculate the gain in thermal energy \( E_c \) of the cold water using the equation

\[ E_c = k(\theta_m - \theta_c) \]

where \( k = 210 \text{ J/}^\circ \text{C} \).

\[ E_c = 210 \times (49 - 23) \]

\[ E_c = 5460 \text{ J} \]

(ii) Calculate the loss in thermal energy \( E_h \) of the hot water using the equation

\[ E_h = k(\theta_h - \theta_m) \]

where \( k = 210 \text{ J/}^\circ \text{C} \).

\[ E_h = 210 \times (83 - 49) \]

\[ E_h = 7140 \text{ J} \] [2]
(c) The student suggests that all the thermal energy lost by the hot water is gained by the cold. Thus $E_c$ and $E_h$ should be equal.

(i) State whether the experimental results support this suggestion. Justify your statement by reference to the results.

<table>
<thead>
<tr>
<th>statement</th>
<th>No</th>
</tr>
</thead>
<tbody>
<tr>
<td>justification</td>
<td>difference is too large</td>
</tr>
</tbody>
</table>

(ii) Suggest a practical reason in this experiment why $E_c$ might be different from $E_h$.

| heat loss to surroundings | heat gained by container (or any sensible suggestion) |

(d) Another student is asked to suggest quantities that should be kept constant if this experiment is repeated in order to check the readings. Table 2.1 shows the suggestions.

Place a tick (✓) in the second column of the table next to each correctly suggested quantity.

<table>
<thead>
<tr>
<th>suggested quantities</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>avoid parallax (line of sight) errors when taking readings</td>
<td></td>
</tr>
<tr>
<td>number of stirs</td>
<td></td>
</tr>
<tr>
<td>room temperature</td>
<td>✓</td>
</tr>
<tr>
<td>starting temperature of hot water</td>
<td>✓</td>
</tr>
<tr>
<td>use a digital thermometer</td>
<td></td>
</tr>
<tr>
<td>use only two or three significant figures for the final answers</td>
<td></td>
</tr>
</tbody>
</table>

-1 for extra ticks in boxes 1, 2, 5 or 6

Note: - if only two boxes ticked, 1 correct & 1 incorrect scores 1 mark

[Total: 7]
3 The IGCSE class is investigating the resistance of a wire.

The circuit is shown in Fig. 3.1.

![Circuit Diagram](image)

Fig. 3.1

(a) A student measures and records in Table 3.1 the current $I$ in the circuit and the potential difference $V$ across a length $l = 0.250$ m of wire PQ.

She repeats the procedure using $l$ values of 0.500 m and 0.750 m.

(i) Complete the heading for each column of the table.

(ii) Calculate the resistance $R$ of each length $l$ of the wire using the equation $R = \frac{V}{I}$.

Record the values of $R$ in the table.

<table>
<thead>
<tr>
<th>$l$ (m)</th>
<th>$I$ (A)</th>
<th>$V$ (V)</th>
<th>$R$ (Ω)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.250</td>
<td>0.54</td>
<td>0.32</td>
<td>1.6875</td>
</tr>
<tr>
<td>0.500</td>
<td>1.10</td>
<td>0.32</td>
<td>3.4375</td>
</tr>
<tr>
<td>0.750</td>
<td>1.61</td>
<td>0.32</td>
<td>5.03125</td>
</tr>
</tbody>
</table>

2 or more s.f.
(b) Use numbers from the table to suggest and justify a relationship between the length \( l \) of the wire and its resistance \( R \). Show your working.

\[ R \text{ is directly proportional to } l \]

justification

\[ l \text{ inc from } 0.250 \text{ to } 0.500 \]

\[ R \text{ inc from } 1.6875 \text{ to } 3.4375 \]

(c) Use the results to predict the resistance of a 1.50 m length of the same wire. Show your working.

by adding the length value to 1.5 m

so \( Rs \) will be added to \( \times 10 \)

or \( \times \) or ratio or proportion prediction \( 10 \rightarrow 10.35 \)

(d) Another student proposes that the accuracy of the experiment would be improved by using a 12V power source.

Suggest two effects that this might have on the experiment.

1. wire gets hot / burn out
   meter damaged

2. wire gets floppy / expands
   higher meter readings / off scale

   power source cuts out
   fuses cut out
   resistance of wire increases

(any two...)

[Total: 11]
The IGCSE class is investigating the formation of images by a lens.

Fig 4.1 shows the apparatus.

![Fig. 4.1](image)

A student places the screen about 1.0 m from the illuminated object. He places the lens between the object and the screen at a distance \( u = 0.200 \) m from the object. He adjusts the position of the screen until a clearly focused image is formed on the screen. He records the distance \( v \) between the centre of the lens and the screen. He repeats the procedure using different values of \( u \). The readings are shown in Table 4.1.

<table>
<thead>
<tr>
<th>( u/\text{m} )</th>
<th>( v/\text{m} )</th>
<th>( 1/1/v/\text{m} )</th>
<th>( 1/1/\text{u/\text{m}} )</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.200</td>
<td>0.596</td>
<td>5.00</td>
<td>1.68</td>
</tr>
<tr>
<td>0.300</td>
<td>0.304</td>
<td>3.33</td>
<td>3.29</td>
</tr>
<tr>
<td>0.400</td>
<td>0.244</td>
<td>2.50</td>
<td>4.10</td>
</tr>
<tr>
<td>0.500</td>
<td>0.214</td>
<td>2.00</td>
<td>4.67</td>
</tr>
<tr>
<td>0.600</td>
<td>0.198</td>
<td>1.67</td>
<td>5.05</td>
</tr>
</tbody>
</table>

(a) State and briefly explain one precaution you would take in order to obtain reliable measurements in this experiment.

precaution: Use of darkened room

explanation: To allow image on screen to be clear enough

or any one:
- how to avoid parallax when taking readings
- moving lens back and forth to obtain clearest image
- mark at centre of lens holder
- place / secure ruler on the bench
- lens, object, screen perpendicular to the bench.
(b) Plot the graph of $\frac{1}{\sqrt{\frac{1}{m}}}$ (y-axis) against $\frac{1}{u} \frac{1}{m}$ (x-axis). Both axes must start at 0 and extend to 7.0.

(c) (i) Use the graph to find the intercept on the y-axis.

intercept on the y-axis = 6.4 \rightarrow 7.0

(ii) Use the graph to find the intercept on the x-axis.

intercept on the x-axis = 6.4 \rightarrow 7.0

[Total: 7]
5  The IGSCE class is determining the density of modelling clay by two methods.

(a) Method 1

A student moulds a piece of modelling clay into a cube shape as shown in Fig. 5.1.

![Fig. 5.1](image)

On Fig 5.1, measure the height \(h\), width \(w\) and depth \(d\) of the cube-shaped piece of modelling clay.

(i) \(h = 3.6\) cm

(ii) \(w = 3.4\) cm

(iii) \(d = 3.2\) cm

(ii) Calculate the volume \(V\) of the modelling clay using the equation \(V = hwd\).

\[
V = 3.6 \times 3.4 \times 3.2
\]

\(V = 39.168\) cm\(^3\)

(iii) Calculate the density \(\rho\) of the modelling clay using the equation \(\rho = \frac{m}{V}\), where the mass of the modelling clay \(m = 103\) g.

\[
\rho = \frac{m}{V} = \frac{103}{39.168}
\]

\(\rho = 2.64\) g/cm\(^3\) [3]

or \(2.6\)

or \(2.63\)
(b) Method 2

The student cuts the piece of modelling clay into two pieces. One piece is approximately twice the size of the other piece. The mass $m_s$ of the smaller piece is 34.5 g.

Fig. 5.2a shows a measuring cylinder containing water. Fig. 5.2b shows the same measuring cylinder after the smaller piece of modelling clay has been lowered into it.

(i) Record the volume of water $V_1$ in the measuring cylinder, as shown in Fig. 5.2a.

$$V_1 = 50 \text{ cm}^3 \quad [1]$$

(ii) Record the new volume $V_2$ in the measuring cylinder, as shown in Fig. 5.2b.

$$V_2 = 64 \text{ cm}^3 \quad [1]$$

(iii) Describe briefly one precaution you would take to read the measuring cylinder correctly.

```
take readings from
Bottom of meniscus
```

(iv) Calculate the volume $V_s$ of the modelling clay using the equation $V_s = (V_2 - V_1)$.

$$V_s = V_2 - V_1 = 64 - 50 = 14 \text{ cm}^3$$

(v) Calculate the density $\rho$ of the modelling clay using the equation $\rho = \frac{m_s}{V_s}$, where $m_s = 34.5$ g.

$$\rho = \frac{34.5}{14} = 2.46 \text{ g/cm}^3 \quad [1]$$

Both are needed.

[Turn over]
(c) (i) Assuming that the experiment has been carried out with care, suggest two reasons why the two values obtained for the density of the modelling clay in (a) and (b) may not be the same.

1. Difficulty of making perfect cuboid shape
   or smaller mass so greater inaccuracy
2. or air bubbles in clay / uneven density distribution
   clay may stick to the knife or any

(ii) State which of the two methods for determining density (method 1 or method 2) you judge to be less accurate. Give a reason for your judgement.

<table>
<thead>
<tr>
<th>method</th>
<th>Method 1</th>
<th>Method 2</th>
</tr>
</thead>
<tbody>
<tr>
<td>reason</td>
<td>sensible matching reason</td>
<td></td>
</tr>
</tbody>
</table>

[Total: 10]